STUDENT'S WORKSHEET.

LESSON 1.

Warm up.

Electricity is pervasively present in our daily lives. We shall begin our study of electricity by studying Electrostatics. Suggest some words you believe can be related to this subject.

Activity 1. You will see some experiments on electrostatics.

task: Try to find what questions they raise. Discuss them with your teacher and the other students in your class.

<u>Activity 2</u>. Watch the video on conductors and insulators your teacher will show you and do the following tasks.

task 1 Write down all the verbs which describe the charged particles' actions. For ex: move around, are stuck, ...

task 2 Identify the main differences between an insulator and a conductor and, in small groups of three, write a paragraph using at least 6 of the verbs from task 1.

task 3 Work in pairs, Discuss in detail how the video can explain the experimental results and answer the following questions. Then Read your answers and discuss them with your class mates.

- 1. What happened when I rubbed the plastic bar?
- 2. Why did the leaves of the electroscope diverge when I touched them with the charged bar?

3. Why could I not charge the metal bar?

Activity 3/ HOMEWORK : Access the page <u>http://education.jlab.org/reading/electrostatics.html</u> and do the exercise. Don't forget to either save / download your exercise or write down your answers in your notebook.

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LESSON 2.

Activity 1:

task 1 (3 min.) Answer your teacher's questions and try to say what questions /problems still have to be tackled.

task 2: (5 min.) Watch the video at

http://www.ck12.org/physics/Electrostatics/lecture/Electrostatics/?referrer=featured_content

Activity 3: (10 min.)

Read the following summary on electrostatics, find the wrong words and replace them with the correct ones (10 minutes).

(Courtesy of Zanichelli www.zanichelli.it)



atoms and molecules

Only within the past century has it become clear that an understanding of electricity originates from within the atom itself. This simplified model of an atom shows a small,dense, negatively charged nucleus containing a mixture of protons and electrons surrounded by

negatively charged neutrons.



All protons and electrons have exactly the same magnitude of electric charge.

Charge on the electron:

e = −1.602× 10⁻¹⁹ C

In the SI, the unit of electric charge is the ampere. Electric charge is quantized in units of electron charge. The total charge carried by any object is a whole multiple of the electron charge.



insulators and conductors

Conductors Charge flows freely: plastics



The relative magnitude of conductivity between silver (a good conductor) and rubber (a good insulator) is on the order of 10²¹.

charging methods



away from the external positive charge.

electroscope

An electroscope is a device used to detect electric current. It is made of two wooden leaves that are free to move, which are situated inside an insulated case.





Activity 4: (10 min.)

Now answer the questions at <u>http://www.ck12.org/physics/Electrostatics/asmtpractice/Electrostatics-Practice/?</u> referrer=featured_content

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LESSON 3.

Activity 1.

- In our approach to electrostatics there is something which we have left unexplained. Can you say what?
- Watch the video at <u>https://www.youtube.com/watch?v=x1-SibwIPM4</u>
- Any questions?

Activity 2.

The electrostatic force has been investigated by the French physicist Charles-Augustin de Coulomb in the 1780s. He found that the module of the force between two point-like charges q_1 and q_2 at a

distance *d* in vacuum is

$$F = \frac{1}{4\pi\epsilon_0} \frac{q_1 q_2}{d^2}$$

We shall now investigate the meaning of that ε_0 .

Activity 3.

Look at the following table. Discuss in pairs how it explains the observation that salt (NaCl) dissolves in water. (3 min.)

Relative permittivities of some	materials at room	temperature
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Material	ε _r	
Vacuum	1 (by definition)	
Air	$1.00058986 \pm 0.00000050$	
PTFE/Teflon	2.1	
Polyethylene/XLPE	2.25	
<u>Polyimide</u>	3.4	
Paper	3.85	
Concrete	4.5	
Pyrex (Glass)	4.7 (3.7–10)	
<u>Diamond</u>	5.5–10	
<u>Salt</u>	3–15	
<u>Graphite</u>	10–15	
Silicon	11.68	
Methanol	30	
Ethylene glycol	37	
Water	88, 80.1, 55.3, 34.5 (0, 20, 100, 200°C)	
ydrofluoric acid	175, 134, 111, 83.6 (-73°C, -42°C, -27°C, 0°C)	
Formamide	84.0 (20°C)	

Activity 4: assessment.

- 1. What happens when a glass bar is positively charged by rubbing it with some piece of cloth?
 - a) Some electrons jump from the cloth to the bar
 - b) Some electrons jump from the bar to the cloth
 - c) Some protons jump from the cloth to the bar
 - d) Both protons jump from the cloth to the bar and electrons from the bar to the cloth
- 2. You want to charge by induction a neutral electroscope using a charged body. Here are the actions you can perform, written in a casual order:
 - a) _____ remove the connection of the electroscope from the ground
 - b) _____ touch the electroscope with the charged body
 - c) ____ connect the electroscope to the ground
 - d) ____ bring the charged body close to the electroscope, without touching it
 - e) ____ turn the charged body away from the electroscope

fill in the blank spaces with the number indicating the correct order of the action; pay attention: one of the actions must not be done: leave the space empty!

3. Fill in the blanks

_____ charges attract;

 charges	

4. Why is it not possible to charge an insulator by induction?

5. Where do the excess electrons end up in a negatively charged conductor, and why?